ORGANIC LETTERS

2012 Vol. 14, No. 6 1504–1507

Abibalsamins A and B, Two New Tetraterpenoids from *Abies balsamea* Oleoresin

Serge Lavoie,[†] Jean Legault,[†] Charles Gauthier,^{†,‡} Vakhtang Mshvildadze,[†] Sylvain Mercier.[†] and André Pichette*,[†]

Université du Québec à Chicoutimi, Chaire de Recherche sur les Agents Anticancéreux d'Origine Naturelle, Laboratoire d'Analyse et de Séparation des Essences Végétales (LASEVE), Département des Sciences Fondamentales, 555 boul. de l'Université, Chicoutimi (Québec), Canada, G7H 2B1, and Université de Poitiers, Institut de Chimie IC2MP, UMR-CNRS 7285, 4 rue Michel Brunet, 86022 Poitiers, France

andre pichette@uqac.ca

Received January 30, 2012

ABSTRACT

RO₂C
$$\tilde{C}$$
O₂H \tilde{C} O₂

Abibalsamins A (1) and B (2), two unprecedented tetraterpenoids featuring a 3,4-seco-rearranged lanostane system fused with a β -myrcene lateral chain *via* a [4 + 2] Diels—Alder cycloaddition, were isolated from the oleoresin of *Abies balsamea*. Their structures were elucidated by means of extensive 2D NMR, IR, and MS spectroscopy analyses. The absolute configuration of 1 was determined by single-crystal X-ray diffraction. Both compounds exhibited significant cytotoxic activity against cancer cell lines.

The genus *Abies* (Pinaceae) consists of 46 species, which are mainly distributed in temperate and boreal regions of North and Central America, Europe, Asia, and North Africa. Previous investigations on the chemical composition of firs (*Abies*) led to the identification of several secondary metabolites such as triterpenoids, flavonoids, stilbenes, chalcones, and lignans, some of them exhibiting pharmaceutically relevant biological activities. ^{2–4}

Balsam fir *Abies balsamea* (L.) Mill. is widely distributed in the eastern part of Canada. Traditionally, it has been

used as an antiseptic, tuberculosis remedy, and venereal aid.⁵ Balsam fir, like other true firs, is characterized by the presence of blisters on the surface of young bark that contain an aromatic liquid called the cortical oleoresin. Oleoresin is known as a complex mixture of mono-, sesqui-, di-, and triterpenoids.^{6–9} Aboriginal people in Canada have employed fir oleoresin as a vulnerary, although nowadays its main field of utilization is as a cement for lenses and as mounting medium in microscopy.¹⁰

In the course of our research program aiming at the isolation and identification of bioactive substances from

[†] Université du Québec à Chicoutimi.

[‡]Université de Poitiers.

⁽¹⁾ Mabberley, D. J. *Mabberley's Plant-Book. A Portable Dictionary of Plants, Their Classification and Uses*, 3rd ed.; Cambridge University Press: Cambridge, 2008, p 1.

⁽²⁾ Raldugin, V. A.; Shevtsov, S. A. Chem. Nat. Compd. 1990, 26, 373.

⁽³⁾ Raldugin, V. A.; Shevtsov, S. A. Chem. Nat. Compd. 1991, 26, 373

⁽⁴⁾ Yang, X. W.; Li, S. M.; Shen, Y. H.; Zhang, W. D. Chem. Biodiversity 2008, 5, 56.

⁽⁵⁾ Herrick, J. W.; Snow, D. R. *Iroquois Medical Botany*; Syracuse University Press: 1995.

⁽⁶⁾ Gray, P. S.; Mills, J. S. J. Chem. Soc. 1964, 5822.

⁽⁷⁾ Sutton, B. A.; Woosley, R. S.; Butcher, D. J. Microchem. J. 1997, 56, 332.

⁽⁸⁾ Leibyuk, T. V.; Shmidt, E. N.; Raldugin, V. A. Chem. Nat. Compd. 1990, 26, 651.

⁽⁹⁾ Khan, V. A.; Tkachev, A. V.; Pentegova, V. A. Chem. Nat. Compd. 1988, 24, 606.

⁽¹⁰⁾ Duke, J. A. Handbook of Energy Crops; NewCROPS: 1983.

plant species of Québec's boreal forest, $^{11-13}$ we have become interested in studying the constituents of *A. balsamea*. $^{12,14-16}$ This work led to the isolation of two new tetraterpenoids, abibalsamins A (1) and B (2), which feature an unusual 3,4-seco-rearranged lanostane triterpene fused with a β -myrcene moiety. The closely related abiesonic acid (3) 17 was isolated as well. Here, we report the isolation, structural elucidation, and absolute configuration of the new compounds 1 and 2 (Figure 1) based on spectroscopic data, X-ray crystallographic analysis, and comparison with literature data.

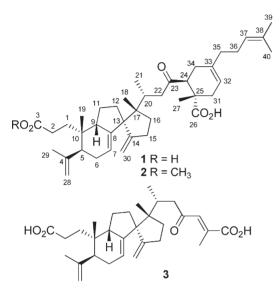


Figure 1. Structures of abibalsamins A (1) and B (2), and the related abiesonic acid (3).

The oleoresin of *A. balsamea* (500 g)¹⁸ was directly subjected to silica gel column chromatography eluting with hexanes/EtOAc as a gradient (100:0 to 93:7) and then MeOH. Both hexanes/EtOAc 93:7 and MeOH fractions were combined and concentrated under reduced pressure (75 g). A portion of the extract (60 g) was further fractionated by silica gel using a gradient of hexanes/EtOAc, affording three subfractions A–C. Subfraction B was purified by silica gel with hexanes/EtOAc (3:1), giving five other subfractions A′–E′. Repeated column chromatography of subfraction C′ on C₁₈ reversed-phase with H₂O/MeOH (1:4

to 0:1) followed by purification using preparative HPLC yielded compounds **1** (44.6 mg) and **2** (13 mg). Crystals of **1** were obtained by recrystallization with EtOH. Purification of subfraction E' on a polyamide column followed by preparative HPLC separation yielded **3** (10 mg), which was identified as abiesonic acid after spectral characterization (IR, NMR, MS).¹⁷ This compound has been thoroughly described in literature in the form of a dimethyl ester.^{8,19}

Compound 1, $[\alpha]_{D}^{20}$ –24.8 (c 0.9, CHCl₃), showed a pseudomolecular ion peak at m/z 619.4348 [M + H]⁺ in the HRESIMS suggesting the molecular formula C₄₀H₅₈O₅. An IR absorption band at 1702 (s) cm⁻¹ implied the presence of carbonyl functionality. Analysis of ¹³C NMR data and HSQC spectrum revealed the presence of seven methyls, twelve sp^3 methylenes, four sp^3 methines, four sp^3 quaternary carbons, two sp^2 methylenes, three sp^2 methines, and eight sp^2 quaternary carbons. A close comparison of the carbon chemical shift of 1 with abiesonic acid (3) suggested a 3,4-seco-rearranged lanostane system with a lateral chain constituted of an additional 6-membered ring. Analysis of the COSY spectrum revealed four spin systems (Figure 2) on the lateral chain which were connected by HMBC cross-peaks from H₃-21 to C-17, C-20, and C-22; from H₂-22 to C-23; from H-24 to C-23; from H₃-27 to C-24, C-25, C-26, and C-31; from H-32 to C-26 and C-35; and from H₃-39 and H₃-40 to C-38 and C-37.

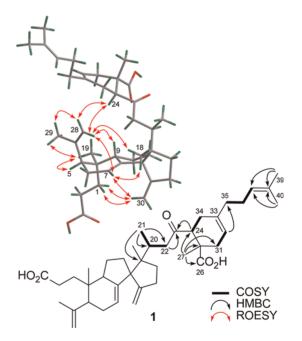


Figure 2. Selected ${}^{1}H-{}^{1}H$ COSY, HMBC, and ROESY correlations for compound **1**.

The relative stereochemistry was established with the ROESY spectrum where cross-peaks from H-28Z to H-9 and from H₃-19 to H-9 and H₃-29 were observed (Figure 2).

Org. Lett., Vol. 14, No. 6, 2012

⁽¹¹⁾ Bradette-Hébert, M. E.; Legault, J.; Lavoie, S.; Pichette, A. Chem. Pharm. Bull. 2008, 56, 82.

⁽¹²⁾ Pichette, A.; Lavoie, S.; Morin, P.; Mshvildadze, V.; Lebrun, M.; Legault, J. Chem. Pharm. Bull. 2006, 54, 1429.

⁽¹³⁾ Simard, F.; Legault, J.; Lavoie, S.; Mshvildadze, V.; Pichette, A. *Phytother. Res.* **2008**, *22*, 919.

⁽¹⁴⁾ Legault, J.; Dahl, W.; Debiton, E.; Pichette, A.; Madelmont, J. C. *Planta Med.* **2003**, *69*, 402.

⁽¹⁵⁾ Pichette, A.; Garneau, F. X.; Collin, G.; Jean, F. I.; Riedl, B. *J. Wood Chem. Technol.* **2003**, *23*, 131.

⁽¹⁶⁾ Pichette, A.; Garneau, F. X.; Jean, F. I.; Riedl, B.; Girard, M. J. Wood Chem. Technol. 1998, 18, 427.

⁽¹⁷⁾ Raldugin, V. A.; Gatilov, Y. V.; Bagryanskaya, I. Y.; Yaroshenko, N. I. Chem. Nat. Compd. 1986, 22, 548.

⁽¹⁸⁾ Oleoresin was recolted by M. Marcel Pichette in summer 2007 at Saguenay, Québec, Canada. A specimen of the scored tree was identified by M. Patrick Nadeau and submitted to herbium Louis-Marie at Université Laval (QFA0579436).

⁽¹⁹⁾ Raldugin, V. A.; Shevtsov, S. A.; Yaroshenko, N. I.; Gatilov, Y. V.; Bagryanskaya, I. Y.; Demenkova, L. I.; Pentegova, V. A. *Chem. Nat. Compd.* **1987**, *23*, 684.

These facts indicated that H-5, Me-19, and H-9 are positioned α , β and β , respectively. Other correlations were observed between H₃-18/H-7 and H-30/H-1 suggesting that ring D is connected in such a way that C-17 is β -oriented and Me-18 is also β . The stereochemistry of the lateral chain could not be established unambiguously with the ROESY spectrum. Single-crystal X-ray diffraction analysis was carried out in order to confirm the structure of 1 (Figure 3). The absolute configuration was determined by Flack's method with Flack's parameter determined as 0.0(2).²⁰ The X-ray structure demonstrated that the chiral centers in 1 were 5S, 9S, 10S, 13R, 17S, 20R, 24S, and 25S. Furthermore, the ROESY correlation between H-28Z and H-24 could be better understood since the measured separation distance was 2.88 Å. On the basis of these spectroscopic evidence, the structure of 1 was assigned as (13R,17S,24S,25S)-24,25-[2-(4-methylpent-3-enyl)but-2-ene-1,4-diyl]-23-oxo-3,4-seco-17,13-friedo-8(14→13)abeo- 9β H-lanost-4(28),7,14(30)-triene-3,26-dioic acid and named abibalsamin A.21

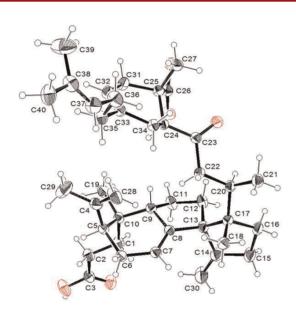


Figure 3. X-ray structure of abibalsamin A (1).

Compound **2**, $[\alpha]^{20}_{D}$ –44.5 (*c* 1.0, CHCl₃), showed a pseudomolecular ion peak at m/z 633.4514 [M + H]⁺ in the HRESIMS suggesting the molecular formula $C_{41}H_{60}O_{5}$. IR absorption bands at 1737 and 1708 (s) cm⁻¹ implied the presence of carbonyl functionalities. Comparison of the ¹H and ¹³C NMR spectra with those of compound **1** revealed that they differ only by an additional methyl group linked at one of the carboxylic functionalities (Table 1). An upfield shift at C-3 (6.4 ppm) for **2** along with an HMBC correlation between the new OCH₃ group and C-3 was observed and

Table 1. NMR Data of Compounds 1 and 2 (δ in ppm, J in Hz)

	1		2	
no.	$\delta_{ m H}$	$\delta_{ m C}$	$\delta_{ m H}$	$\delta_{ m C}$
1	1.68, m (2H)	29.7	1.63, m	30.7
			1.76, m	
2	2.34, m (2H)	28.7	2.31, m (2H)	29.3
3	_	181.3	_	174.9
4	_	149.4	_	149.3
5	2.05, m	43.9	2.07, m	44.0
6	2.14, m	31.1	2.14, m	31.0
	2.34, m		2.39, m	
7	5.43, br s	122.2	5.45, br s	122.1
8	_	144.0	_	143.7
9	2.15, m	49.2	2.17, m	49.2
10	_	37.3	_	37.0
11	1.34, m	22.4	1.39, m	22.5
	1.67, m		1.64, m	
12	1.26, m	31.3	1.31, m	31.2
	1.66, m		1.75, m	
13	_	63.6	_	63.5
14	_	161.7	_	161.5
15	2.35, m	28.0	2.36, m	27.9
	2.47, m		2.47, m	
16	1.54, m (2H)	36.5	1.56, m (2H)	36.4
17	_	50.5	_	50.4
18	0.87, s	18.0	0.89, s	18.0
19	0.93, s	25.0	0.94, s	24.9
20	2.31, m	33.1	2.33, m	33.9
21	0.81, d (6.5)	17.0	0.80, d (6.4)	16.6
22	2.12, m	44.6	2.14, m	44.7
	2.36, m	11.0	2.30, m	11
23	_	211.5	_	211.4
24	3.12, dd (12.4, 5.1)	50.4	3.08, dd (11.8, 5.4)	50.4
25	-	42.5	-	42.3
26	_	185.0	_	183.6
27	1.29, s	15.7	1.25, s	16.1
28	Z 4.92, br s	111.7	4.83, s (2H)	111.9
20	E 4.82, s	111.1	1.00, 5 (211)	111.0
29	1.78, s	26.5	1.77, s	26.3
30	E 4.76, s	106.8	E 4.76, s	106.7
	Z 4.67, br s		Z 4.72, br s	
31	2.01, m	38.1	2.05, m	37.6
01	2.28, m	50.1	2.33, m	01.0
32	5.39, d (3.8)	118.8	5.39, br s	118.9
33	- -	135.8	-	135.4
34	1.87, m	29.0	1.87, m	28.7
01	2.22, m	20.0	2.27, m	20.1
35	2.06, m (2H)	37.1	2.03, m	37.1
99	2.00, iii (211)	01.1	2.08, m	01.1
36	2.10, m (2H)	26.3	2.09, m (2H)	26.3
37	5.07, t (6.7)	123.7	5.06, t (5.9)	123.8
38	_	131.9	_	131.9
39	1.68, br s	25.7	- 1.68, br s	25.7
40	1.60, br s 1.61, br s	17.8	1.60, br s	17.8
Me			3.67, s	51.6
1416		_	0.01, 5	01.0

allowed us to assign the structure of **2**, named abibalsamin B, ²² as (13R,17S,24S,25S)-24,25-[2-(4-methylpent-3-enyl)-but-2-ene-1,4-diyl]-23-oxo-3,4-seco-17,13-friedo-8(14 \rightarrow 13)-abeo-9 β H-lanost-4(28),7,14(30)-diene-3,26-dioic acid 3-methyl ester.

Although some marine species have been shown to contain polycyclic tetraterpenoids, $^{23-25}$ the isolation of noncarotenoid C_{40} compounds from plants is very scarce. 26

1506 Org. Lett., Vol. 14, No. 6, 2012

⁽²⁰⁾ Flack, H. D. *Acta Crystallogr*. **1983**, *A39*, 876. (21) Abibalsamin A (1): white crystals (EtOH), $[\alpha]^{20}_{\rm D}$ –24.8 (*c* 0.9, CHCl₃); IR (film) $\nu_{\rm max}$ 2960, 1742, 1702, 1460, 1376, 1295, 1217, 1160, 897, 756 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) and ¹³C NMR (100 MHz, CDCl₃) data, see Table 1; HR-ESI-MS m/z 619.4348 [M + H]⁺ (calcd for C₄₀H₅₉O₅, 619.4357).

Scheme 1. Plausible Biosynthetic Pathway of 1 and 2

A plausible biosynthetic pathway of abibalsamins A (1) and B (2) is proposed in Scheme 1. In opposition to 'true' tetraterpenes such as the carotenoids, which are formed by a head-to-head condensation of two geranylgeranyl pyrophosphate (GGPP) units,²⁷ it is likely that abibalsamins 1 and 2 would come from a different biosynthetic route. Since abiesonic acid (3), a 3,4-seco-rearranged lanostane-type triterpenoid,²⁸ and the monoterpene β -myrcene have been isolated from *A. balsamea*,²⁹ it seems reasonable to propose that abibalsamin A (1) would be formed by an

enzyme-catalyzed [4 + 2] Diels—Alder-like cycloaddition between 3 and β -myrcene. Abibalsamin B (2) would then be obtained by the esterification of the C-3 carboxylic acid function in 1. Such a type of Diels—Alder-like biosynthetic reaction has already been suggested for dimeric triterpenoids³⁰ and tricyclic spirolactones,³¹ to name a few recent examples.

Table 2. Cytotoxic Activity of Compounds $1-3^a$

	Compounds				
Cell lines	1	2	3	Etoposide	
A549	22 ± 4	8.5 ± 0.7	22 ± 3	0.3 ± 0.1	
DLD-1	>100	15 ± 1	>100	1.0 ± 0.4	
WS1	>100	14.7 ± 0.2	30 ± 2	5 ± 1	

The *in vitro* antiproliferative activities of compounds **1–3** against lung carcinoma (A549), colorectal adenocarcinoma (DLD-1), and normal skin fibroblast (WS1) human cell lines were evaluated using the resazurin reduction test as previously described. The cytotoxicity results presented in Table 2 are expressed as the concentration inhibiting 50% of the cell growth (IC50). Etoposide was used as a positive control in this assay (IC50 0.3–5 μ M). The most cytotoxic compound was abibalsamin B (2) with IC50 values of 8.5 \pm 0.7 and 15 \pm 1 μ M against A549 and DLD-1 cancer cell lines, respectively. Interestingly, abibalsamin A (1) selectively inhibits the growth of A549 cells (IC50 22 \pm 4 μ M) compared to DLD-1 and WS1 (IC50 > 100 μ M).

Acknowledgment. The authors acknowledge the "Chaire de recherche sur les agents anticancéreux d'origine naturelle" and NSERC for funding.

Supporting Information Available. Experimental section, NMR and IR spectra of new compounds (1 and 2). This material is available free of charge via the Internet at http://pubs.acs.org.

The authors declare no competing financial interest.

Org. Lett., Vol. 14, No. 6, 2012

⁽²²⁾ Abibalsamin B (2): white amorphous solid, $[\alpha]^{20}_{D}$ –44.5 (c 1.0, CHCl₃); IR (film) ν_{max} 2958, 1737, 1708, 1435, 1375, 1294, 1193, 1172, 896, 756 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) and ¹³C NMR (100 MHz, CDCl₃) data, see Table 1; HR-ESI-MS m/z 633.4514 [M + H]⁺ (calcd for C₄₁H₆₁O₅, 633.4514).

⁽²³⁾ Zeng, L. M.; Lan, W. J.; Su, J. Y.; Zhang, G. W.; Feng, X. L.; Liang, Y. J.; Yang, X. P. J. Nat. Prod. 2004, 67, 1915.

⁽²⁴⁾ Jingyu, S.; Kanghou, L.; Tangsheng, P.; Cun-Heng, H.; Clardy, J. J. Am. Chem. Soc. 1986, 108, 177.

⁽²⁵⁾ Park, S. Y.; Choi, H.; Hwang, H.; Kang, H.; Rho, J.-R. J. Nat. Prod. 2010, 73, 734.

⁽²⁶⁾ Zhang, Y.; Lu, Y.; Mao, L.; Proksch, P.; Lin, W. Org. Lett. 2005, 7, 3037.

⁽²⁷⁾ Hirschberg, J.; Cohen, M.; Harker, M.; Lotan, T.; Mann, V.; Pecker, I. *Pure Appl. Chem.* **1997**, *69*, 2151.

⁽²⁸⁾ Kuroyanagi, M.; Sugiyama, K.; Kanazawa, M.; Kawahara, N. Chem. Pharm. Bull. 2000, 48, 1917.

⁽²⁹⁾ Chung, J. L.; Snajberk, K.; Zavarin, E. Phytochemistry 1974, 13, 179.

⁽³⁰⁾ Hou, X. F.; Yao, S.; Mándi, A.; Kurtán, T.; Tang, C. P.; Ke, C. Q.; Li, X. Q.; Ye, Y. Org. Lett. **2012**, *14*, 460.

⁽³¹⁾ Liu, M.; Lin, S.; Gan, M.; Chen, M.; Li, L.; Wang, S.; Zi, J.; Fan, X.; Liu, Y.; Si, Y.; Yang, Y.; Chen, X.; Shi, J. *Org. Lett.* **2012**, *14*, 1004. (32) Gauthier, C.; Legault, J.; Girard-Lalancette, K.; Mshvildadze, V.; Pichette, A. *Bioorg. Med. Chem.* **2009**, *17*, 2002.